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or more

substances

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a mixture in which particles of a material are more or less ev... in a solution, the substance in which the solute is dissolved.

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12) Multiple
Choice (40%) 1)

Which of the
following is a
kinetic

quantity? A)

Enthalpy B)

Internal Energy

C) Gibb's free
energy D)

Entropy E) Rate

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of reaction 2)
Of the following
questions, which
ones are
thermodynamic,
rather than
kinetic
concepts? I) Can
substances react
when we put them
together?

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12) Multiple Test

Choice (40%)

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Acids. The next

ones in line are

the aldehydes,

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ketones and carboxylic acids. The IUPAC names, to start with, the chapter explores their preparation methods, the key reactions, characteristic tests, and other properties.

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gases. By

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percentage, it
is roughly 78
percent

nitrogen, 21
percent oxygen,
and 1 percent
argon. (There
are trace
amounts of many
other gases in
air.) If the
atmospheric
pressure is 760
mm Hg, what

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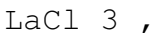
Answer: $4.93 \times$
 10^5 L or 49.3
?L In Example

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12.2.1 and Test

Example 12.2.2,
the identity of
the limiting
reactant has
been apparent:



ethanol, and

para

-nitrophenol.

When the

limiting

reactant is not

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apparent, we can determine which reactant is limiting by comparing the molar amounts of the reactants with their coefficients in the balanced chemical equation, just as we did in Section 11.4 .

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