Chapter 4 Atomic Structure Test B Answers

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Questions : 1. The number of electrons in Na+ (atomic number = 11) are (a) 11 (b) 10 (c) 12 (d) 13 2. There are four elements P, Q, R and S having atomic numbers of 4, 18, 10 and 16 respectively. The element which can exhibit covalency as well as electrovalency will be: (a)...

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The four atomic species can be represented as follows. Out of these, the two species which can be termed isobars are: (i) 201 X 60 (ii) 200 X 61 (iii) 200 X 58 (iv) 203 X 60

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